

# **pharmacognosy**

**3<sup>rd</sup> stage/1<sup>st</sup> term**

## **Glycosides**

**Dr. Zahraa Shubber**

**Lec .3**

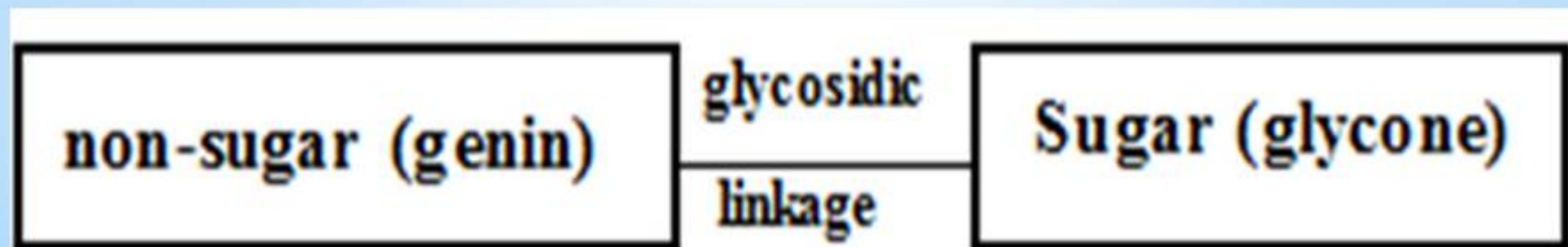


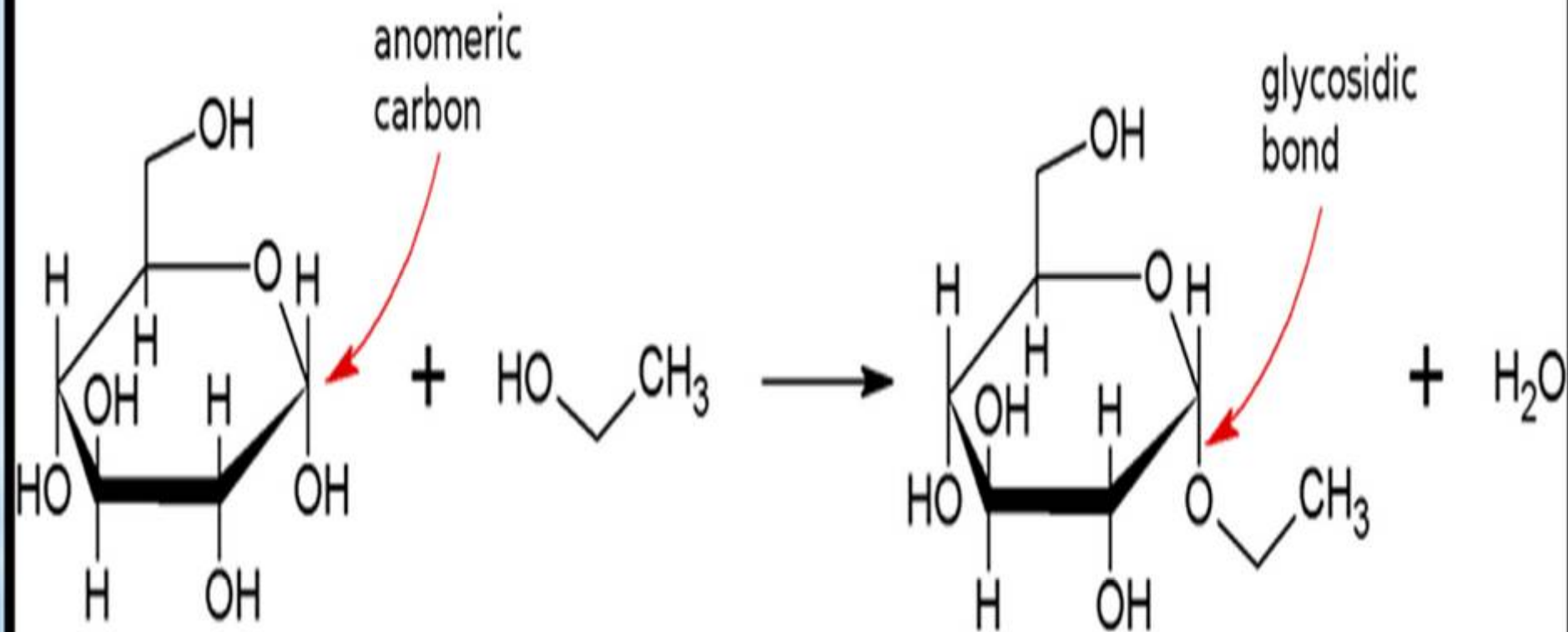
# Glycosides

**Glycosides** are an organic compounds, usually of plant origin, that composed of a **sugar molecule** (such as glucose, fructose, ribose, etc.) and **non-sugar molecule** bonded via a glycosidic bond.

**Glycosides are composed of two portions:**

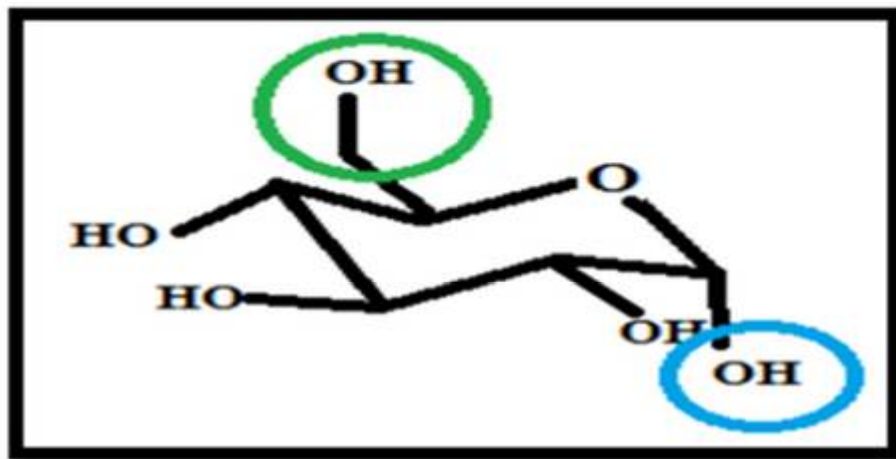
- **Glycone** portion that refer to the sugar part of the glycoside. The sugar component of glycosides may be mono, di, tri or tetra saccharides.
- **Aglycone** (genin) portion that refer to the non-sugar part of glycoside.



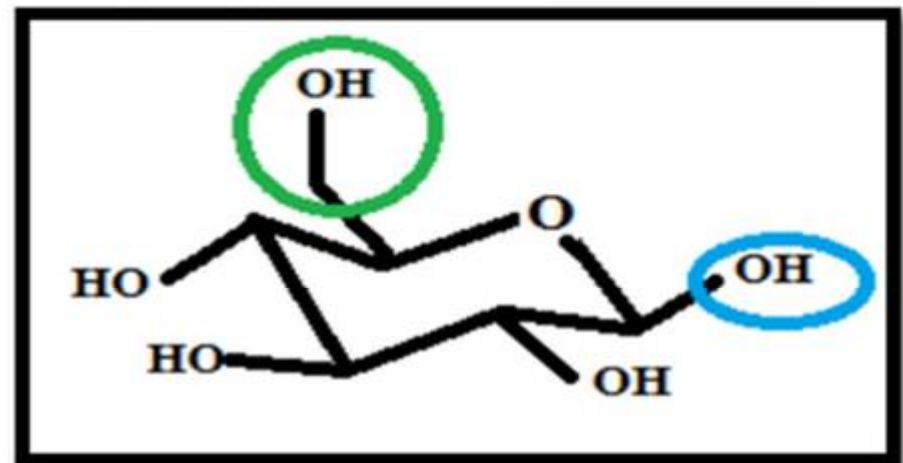


# Alpha and beta glycosides

- Sugars in glycosides exist in isomeric  $\alpha$  and  $\beta$  forms so both  $\alpha$  and  $\beta$  glycosides are theoretically possible.
- The two diastereoisomers differ in configuration about the anomeric carbon (C-1) can exist  $\alpha$  and  $\beta$ .
- If the hydroxyl group on the anomeric carbon is **down** in relation to the cyclic structure, it is  **$\alpha$  anomer** while if the hydroxyl group on the anomeric carbon is **up** in relation to the cyclic structure, it is  **$\beta$  anomer**.



**$\alpha$  anomer**

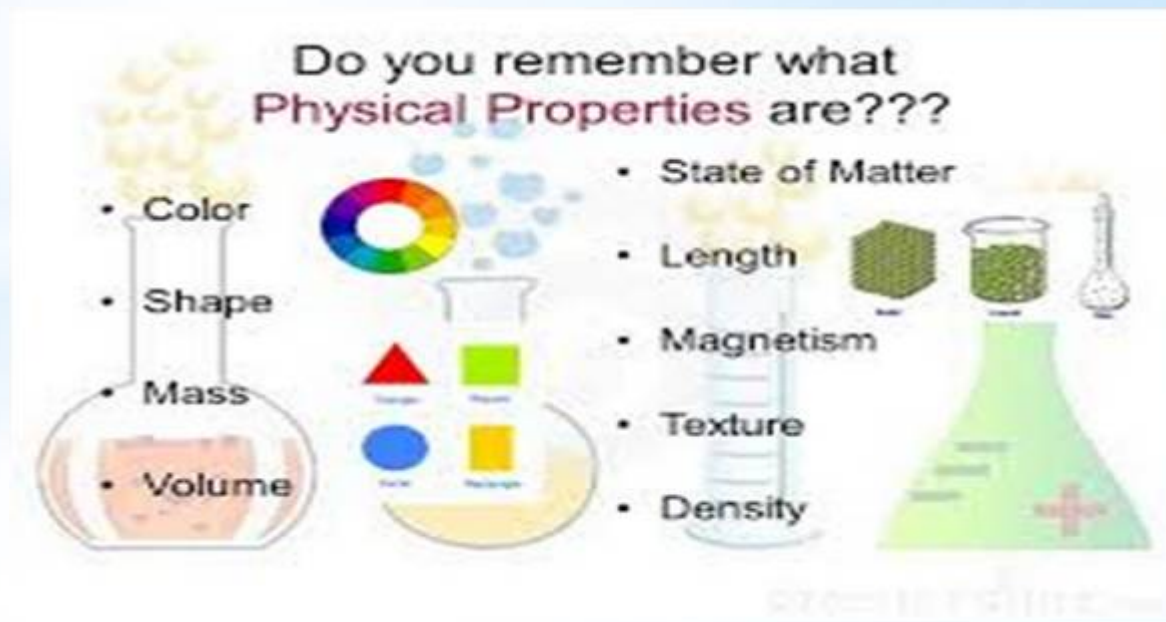


**$\beta$  anomer**

# Physical and chemical properties of glycosides

## 1. Solubility:

- Most glycosides are soluble in water or hydroalcoholic solutions and insoluble or less soluble in non-polar organic solvents, because the solubility properties of the sugar residues exert a considerable effect i.e. sugar moiety increases water solubility.
- The aglycon part is soluble in non-polar (organic) solvents like benzene, ether and chloroform.





## 2. Stability and hydrolytic cleavage:

**A. Acids and alkali:** Glycosides can be hydrolyzed by heating with a dilute acid where by the glycosidic linkages are cleaved, while glycosides are relatively stable towards alkalis.

### **B. Enzyme hydrolysis:**

- Enzymatic hydrolysis is specific for each glycoside there is a specific enzyme that exerts a hydrolytic action on it.
- The same enzyme is capable to hydrolyze different glycosides, but  $\alpha$  and  $\beta$  stereo-isomers of the same glycoside are usually not hydrolyzed by the same enzyme.
- **Emulsin** is found to hydrolysed most  **$\beta$ -glycoside** linkages while **maltase and invertase** are  $\alpha$ -glycosidases, capable of hydrolyzing  **$\alpha$ -glycosides only.**

### 3. Shape, color, taste and odor

**A. Shape:** Glycosides are solid, amorphous and non volatile.

**B. Color:** Glycosides are colourless except flavonoids are yellow and anthraquinones are red or orange. Glycosides give +ve reaction with Fehling's solution test ( after hydrolysis).

**C. Taste:** Most of glycosides are bitter taste.

**D. Odor:** Glycosides are odorless except saponin (glycyrrhizin).

## ***Importance of Glycosides***

- 1. Glycosides play an important role in the life of the plant and are involved in different functions.**

**It serve as:**

- A.**As sugar reserves
- B.**As waste products of plant metabolism
- C.**As a mean of detoxification
- D.** To regulate osmosis
- E.** To regulate the supply of substances of importance in metabolism
- F.** Has a role of defense against the invasion to the tissues by microorganism some pointed out that aglycones are antiseptics and hence are bactericidal in nature



**2. Many therapeutic agents are derived from glycosides. In fact, the group contributes to almost every therapeutic class.**

**A.** Some of our most valuable cardiac glycosides from digitalis, strophanthus, squill and others.

**B.** Laxative drugs, such as senna, aloe, rhubarb, cascara sagrada, and frangula, contain emodin and other anthraquinone glycosides;

**C.** Sinigrin, a glycoside from black mustard, yields allyl isothiocyanate, a powerful local irritant.



Senna



Digitalis

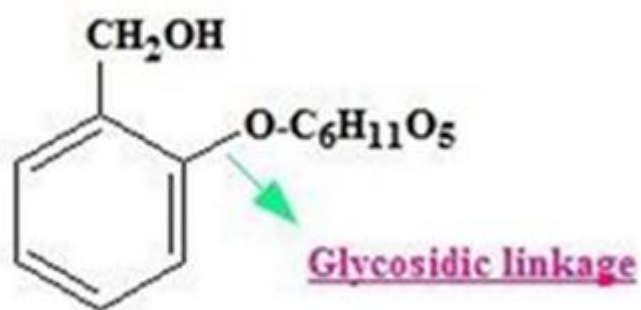
# Classification of Glycosides

## **A. According to the type of glycosidic linkage:**

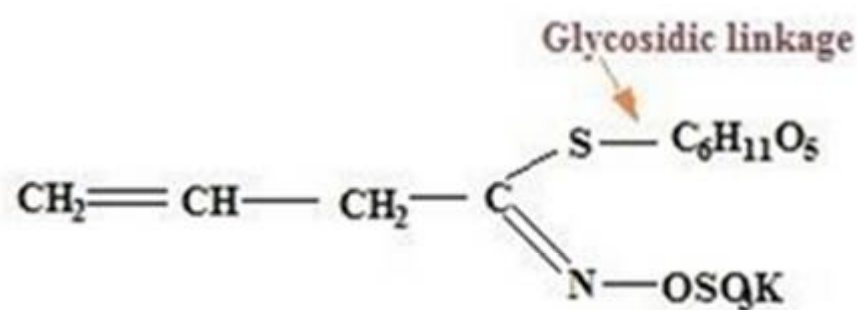
1.  $\alpha$ -glycosides ( $\alpha$  sugar)
2.  $\beta$ -glycosides ( $\beta$  sugar)

## **B. According to the chemical group of the aglycon involved in the formation of glycoside linkage.**

1. **Aglycone-O-Sugar:** O-glycosides (OH-group): eg. Senna and rhubarb
2. **Aglycone-C-Sugar:** C-glycosides (C-group): eg. Cascaroside from cascara
3. **Aglycone-S-Sugar:** S-glycosides (SH-group): eg. Sinigrin from black mustard
4. **Aglycone- N-Sugar:** N-glycosides (NH- group): eg. glycoalkaloid



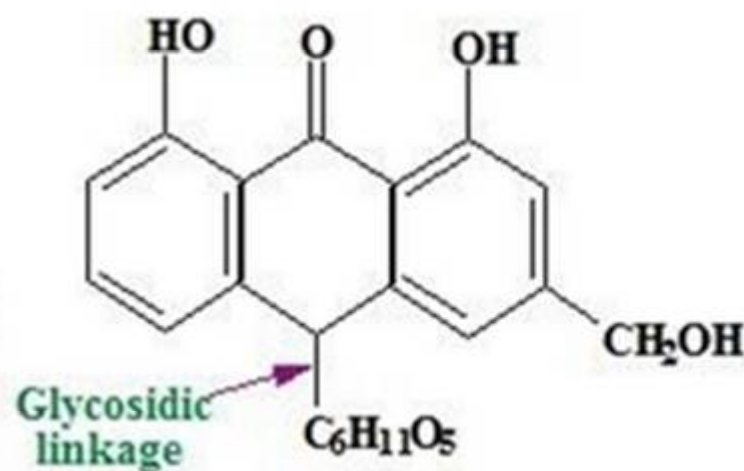
**O-glycoside**



**S-glycoside**



**N-glycoside**



**C-glycoside**

**Types of glycoside based on the Glycosidic bond**

## **C. According to the chemical nature of the aglycon:**

- 1. Cardioactive group**
- 2. Anthraquinone group**
- 3. Saponin group**
- 4. Cyanophore group**
- 5. Isothiocyanate group**
- 6. Flavonol group**
- 7. Alcohol group**
- 8. Aldehyde group**
- 9. Phenol group**

**D. According to the nature of the simple sugar component of the glycoside:**

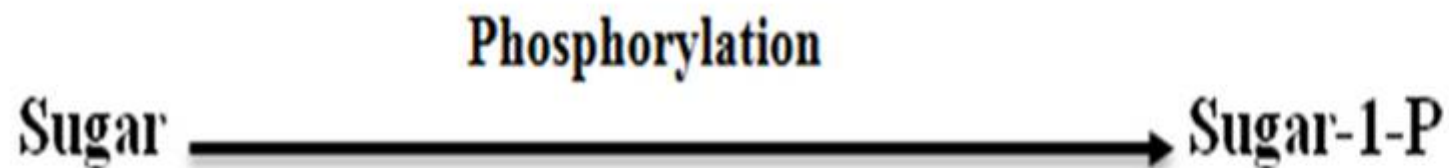
- 1. Glucoside (glycone is glucose)**
- 2. Galactoside (glycone is galactose)**
- 3. Mannoside (glycone is mannose)**
- 4. Arabinoside (glycone is arabinose)**



## ***Biosynthesis of glycosides***

- The biosynthetic pathways are widely variable depending on the type of aglycone as well as the glycone units.
- The aglycone and the sugar parts are biosynthesized separately, and then coupled to form a glycoside.
- The coupling of the two parts occurs via phosphorylation of a sugar to yields a sugar-1-phosphate which reacts with a uridine triphosphate to form a uridine diphosphate sugar (UDP-sugar) and inorganic phosphate.
- This UDP-sugar reacts with the aglycone to form the glycoside and a free UDP.





Glycoside

## Biosynthesis of Glycosides

*Thank you*

