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Types of Dryers

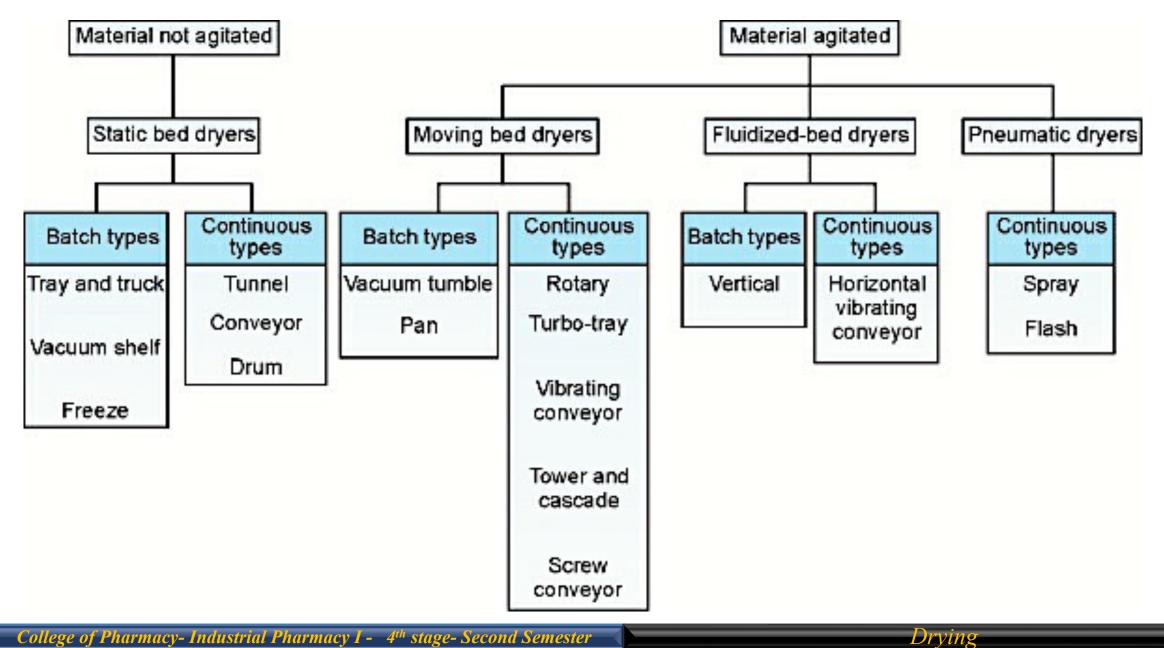
- When considering how to dry material, certain points should be considered:
- 1. Heat sensitivity of the material being dried
- 2. Nature of the liquid to be removed.
- 3. The quantity of product to be dried (scale of operation).
- 4. Physical nature of the material.

- Dryers can be classified based on the method of heat transfer or method of sample handling.
- In the **first** classification **dryer design and energy requirement** are important while in the **second** attention is given to the **type of the substance to be dried**.

Types of Dryer

- Classification by the method of sample handling:
- Two main classes under this classification: presence or absence of agitation of the material to be dried.
 - Agitation dryers are not recommended for friable materials
- Based on this classification there are 4 classes of dryers:
- 1. Static bed dryer
- 2. Moving bed dryer
- 3. Fluidized bed dryer
- 4. Pneumatic dryer

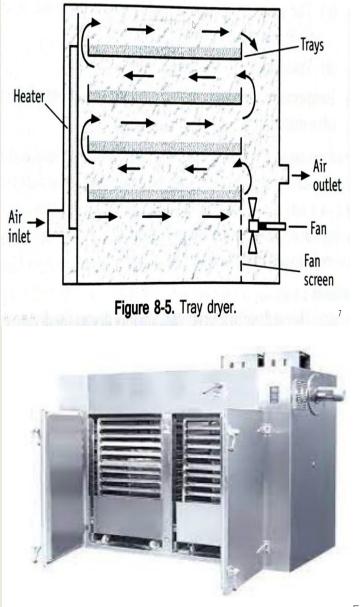
Classification of The Dryers



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Static Bed Dryers: 1- Tray Dryer

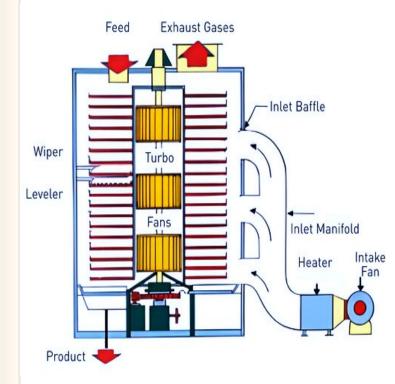
- **The tray dryer** consists of a cabinet shelf or compartment in which the material to be dried is spread on trays.
- There is **no particle movement**; sometimes only movement of the entire drying mass.
- The exposed surface can be **increased by decreasing** the thickness of the bed.
- Advantages:
- 1. Versatile in arrangement and design.
- 2. Relatively **low cost**.



Moving Bed Dryer

Turbo-Tray Dryers

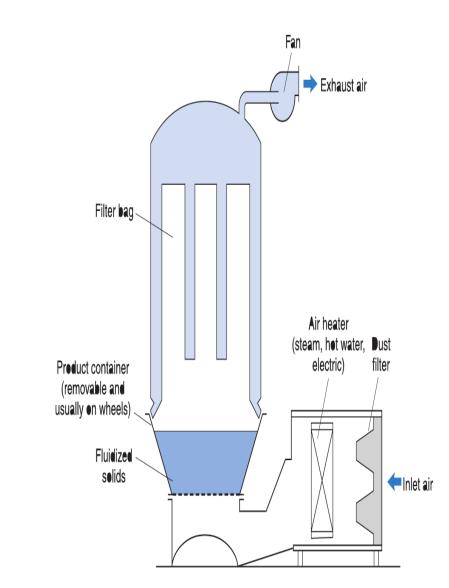
- This type is a **<u>continuous</u> shelf**, moving-bed dryer.
- It consists of a series of rotating annular trays arranged in a vertical stack. Rotating slowly at 0.1 to 1.0 rpm.
- **Heated air** is circulated by turbo-type fans mounted in the stack center.
- Wet mass fed through the roof of the dryer is leveled by a stationary wiper.
- Then the dried material is **pushed through radial slots** onto the tray below.
- After each cycle, the mass transfers to the next shelf until discharge at the bottom.
- The **drying rate** is **faster than the tunnel** dryer due to the continuous exposure to the air.



https://youtu.be/-XpohB2hUoM

Fluidized Bed Dryer (FBD)

- The solids are partially suspended in the gas stream (the mixture behaves like a liquid) and the solid is fluidized.
- The fluidization technique is **used for** drying granular solids, as each particle is surrounded by the drying gas.
- The intense mixing results in uniform conditions of temperature, composition, and particle size distribution throughout the bed.
- The resultant granules are not wet nor completely dried to avoid cracking.



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Advantaged of FBD

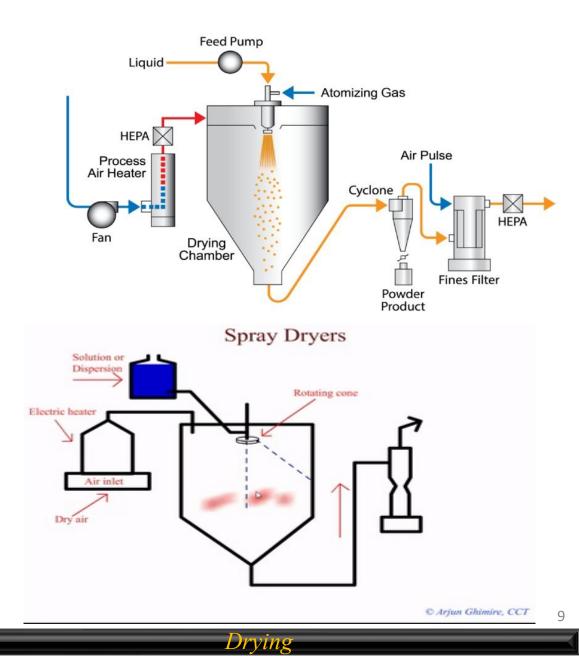
- Efficient heat and mass transfer give high drying rates, so drying times are short.
 - Apart from obvious economic advantages, the heat challenge to thermolabile materials is **minimized**.
- The fluidized state of the bed ensures that drying occurs from the surface of all the individual particles. Hence, most of the drying will occur at a constant rate.
- 3. The turbulence in a fluidized bed causes some attrition to the surface of the granule. This produces a more spherical free-flowing product which will reduce the problem of aggregation and migration of color.



Pneumatic Dryer

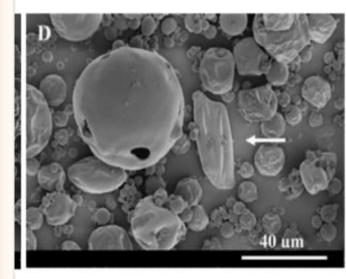
Spray Dryer

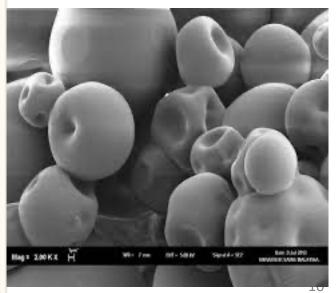
- It can handle **only fluid materials** such as solutions, slurries, and thin pastes.
- Feeding of fine droplets of fluid into the hot gas stream.
- The dried powder is carried by the **gas current and gravity flow** to the collection system.



Principle

- When the liquid droplets come into contact with the hot gas, they quickly reach a high temperature and the water at the surface will evaporate and form a tough shell solid.
- The diffusion rate of the liquid is slower than heat transfer. → The internal pressure causes the droplet to swell, and the shell becomes thinner, allowing faster diffusion.
- **3.** If the shell is nonelastic or impermeable, it ruptures, producing either fragments or bud-like forms.
- Thus, spray-dried material consists of intact spheres, spheres with buds, ruptured hollow spheres, or sphere fragments





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Spray Drying and Spray Congealing

- Spray drying is rapid drying and a unique form of the product.
- There are three major uses:
 - 1. Drying heat-sensitive materials,
 - 2. Changing the physical form of materials (in tablet and capsule manufacture)
 - 3. Encapsulating solid and liquid particles
- Spray drying is used in tablet and capsule formulations.
- The drying process changes the shape, size, and bulk density of the product.
- The spherical particles **flow better** than the same product dried by the conventional method **due to** size and shape uniformity.

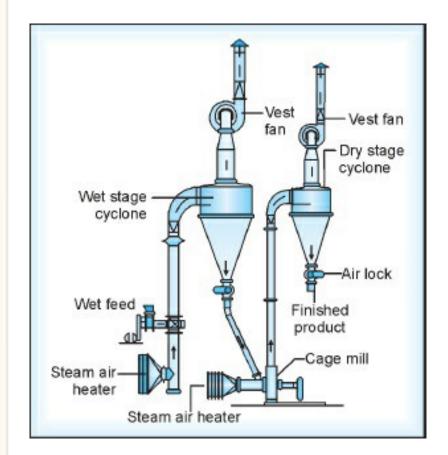
Spray Drying and Spray Congealing

• Spray Congealing:

- Spray drying is used in the coating and encapsulation of both solids and liquids.
- Chilling spray (congealing) consists of suspending the particles in a **molten coating material** and pumping the slurry into a spray dryer in which **cold air** is circulated.
- Spray congealed coatings are **used mainly for 1) taste masking** and 2) **sustained-release** formulations.

Flash Dryer

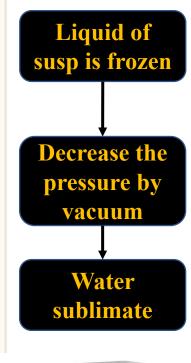
- The moistened solid is suspended in a finely divided state [velocity (3000-6000 feet/min)] at [temperature (300-1300°F] air stream
- The flash drying is a <u>short-time process</u>. This is why it is called a flash dryer.



Specialized Drying Methods

Freeze Drying

- The process is also called lyophilization, and sublimation.
- The drying of **heat-sensitive materials** must be dehydrated to a solid state **to maintain the stability** through frozen then under a high vacuum to heat (by conduction or radiation) to sublime the frozen liquid leaving only the solid.
- Examples; blood serum, plasma, antibiotics, hormones, bacterial cultures, vaccines, and food.
- The dried product is called **lyophilized** and can be redissolved (or resuspended) by adding water before its use by a process called **reconstitution**.
- In freeze-drying the water **passes directly** from the solid state (ice) to the vapor state without passing through the liquid state. As shown in the schematic pressure-temperature diagram for water.

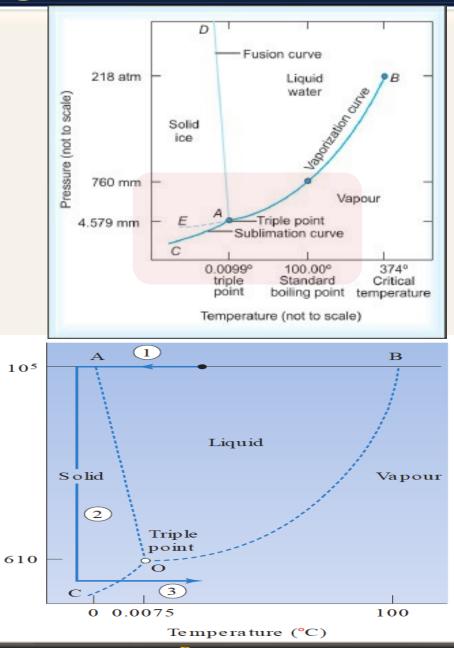




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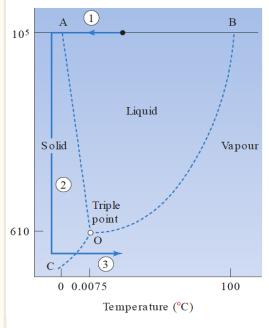
Freeze Drying

- The theory and practice of freeze drying are based on an understanding and application of the **phase diagram** for the water system.
- Sublimation occurs at pressure and temperature below that for the triple point (4.579 mm Hg, and 0.0099 °C)

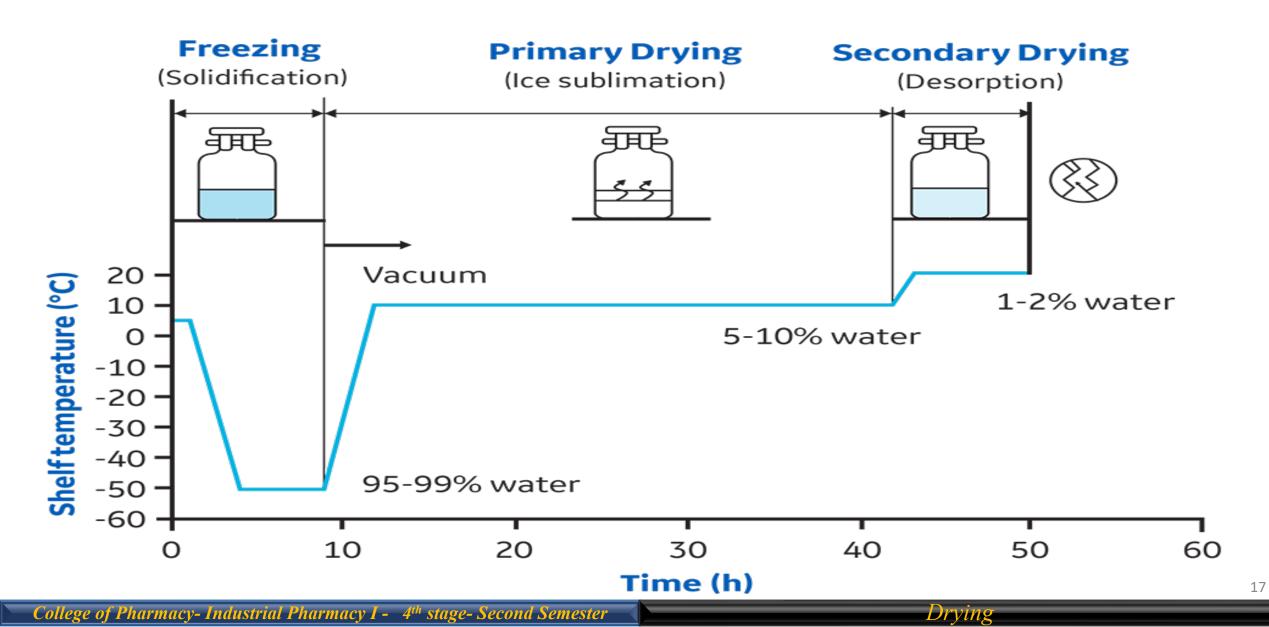


Stages of Freeze Drying

- Pre-freezing: The material is frozen by keeping the material below or at −20°C (freezing before vacuum to prevent foaming).
- 2. Vacuum: Reducing the atmospheric pressure to below that of the triple point of the product
 - Rotary pumps on a small scale and ejector pumps on a large scale are used to reduce the pressure sufficiently.
- Primary Drying: latent heat of sublimation must be provided and the vapor removed. Primary drying by sublimation can remove the unbound water.
 - This stage will result in a **powder with very low moisture** content.
- **4. Secondary Drying**: It is used to remove bound water or traces of water left after primary drying.
 - The temperature is raised (up to 50°C) or desiccant is used to carry secondary drying.

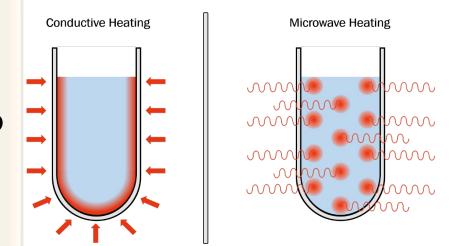


Lyophilization Process

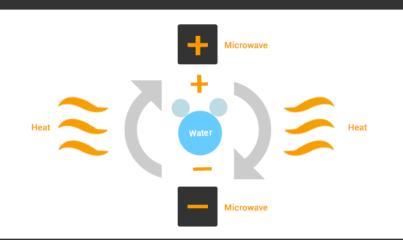


Specialized Drying Methods

- Radiation heat transfer:
- Heat "transmission" by radiation differs from heat transfer by conduction or convection in that no transfer medium (solid, liquid, or gaseous) needs to be present.



Interaction between water molecules and microwave



Microwave Radiation

- The application of microwave energy to the drying of solids represents a radical departure from conventional means of drying.
- Instead of **applying heat externally to a material**, energy in the form of microwaves **is converted** into **internal heat** by interaction with the material itself.
- This permits **extremely rapid** heat transfer throughout the material leading to rapid drying.
- The moisture is mobilized as a vapor rather than a liquid, and its movement to the surface can be extremely rapid because it does not depend on mass concentration gradients or on a slow liquid diffusion rate.

