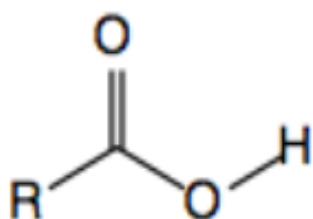
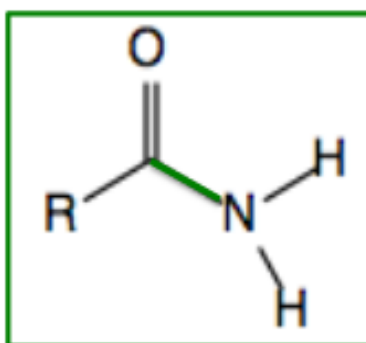


Amides

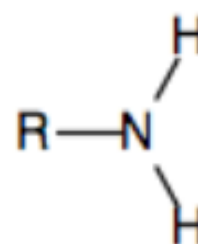
- In **amides**, an amino group ($-\text{NH}_2$) replaces the $-\text{OH}$ group of carboxylic acids.



ethanoic acid



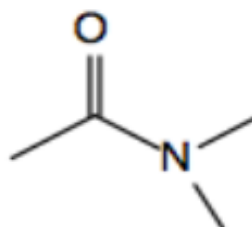
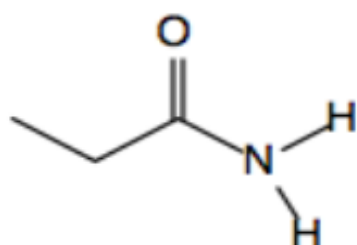
ethanamide



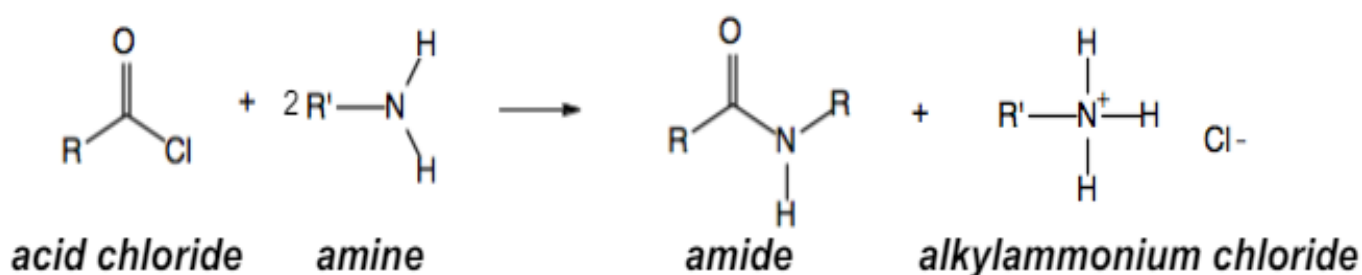
ethanamine

- Amides are (mainly) solids at room temperature and have very high boiling points.
- Simple amides are soluble in water.

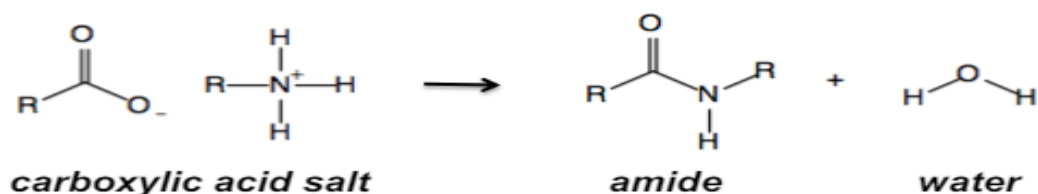
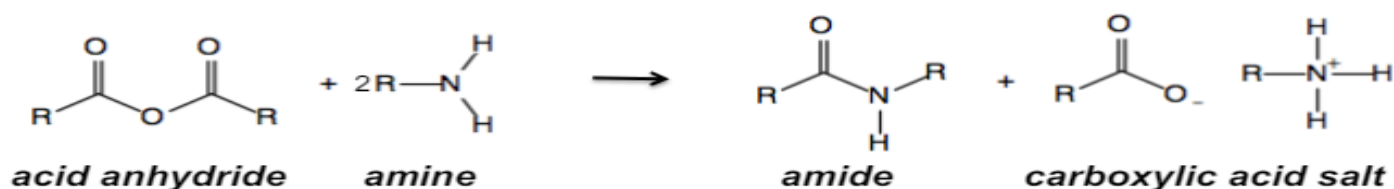
- Name the following compounds.

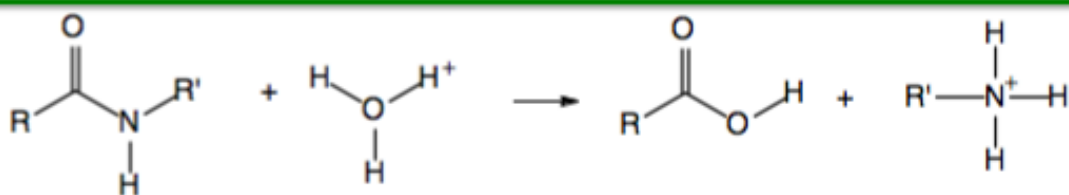


- Primary and secondary amines react with acid chlorides to produce amides.

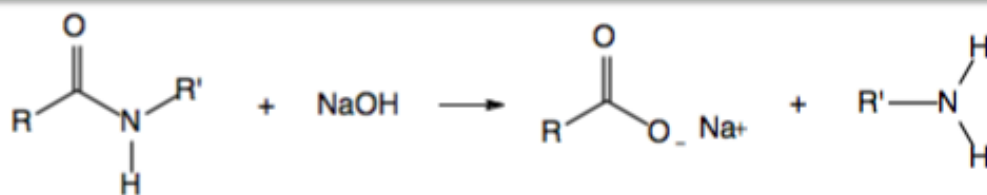


- Primary and secondary amines react with acid anhydrides to produce amides.





In strong acid: carboxylic acid + alkylammonium ion



In strong base: carboxylic acid salt + amine