

Ministry of Higher Education and Scientific Research

Al-Mustaqbal University

College of Medical and Health Technologies

Aesthetic and Laser techniques Department



Laser Physics

Laser level systems

First stage

Lecture4

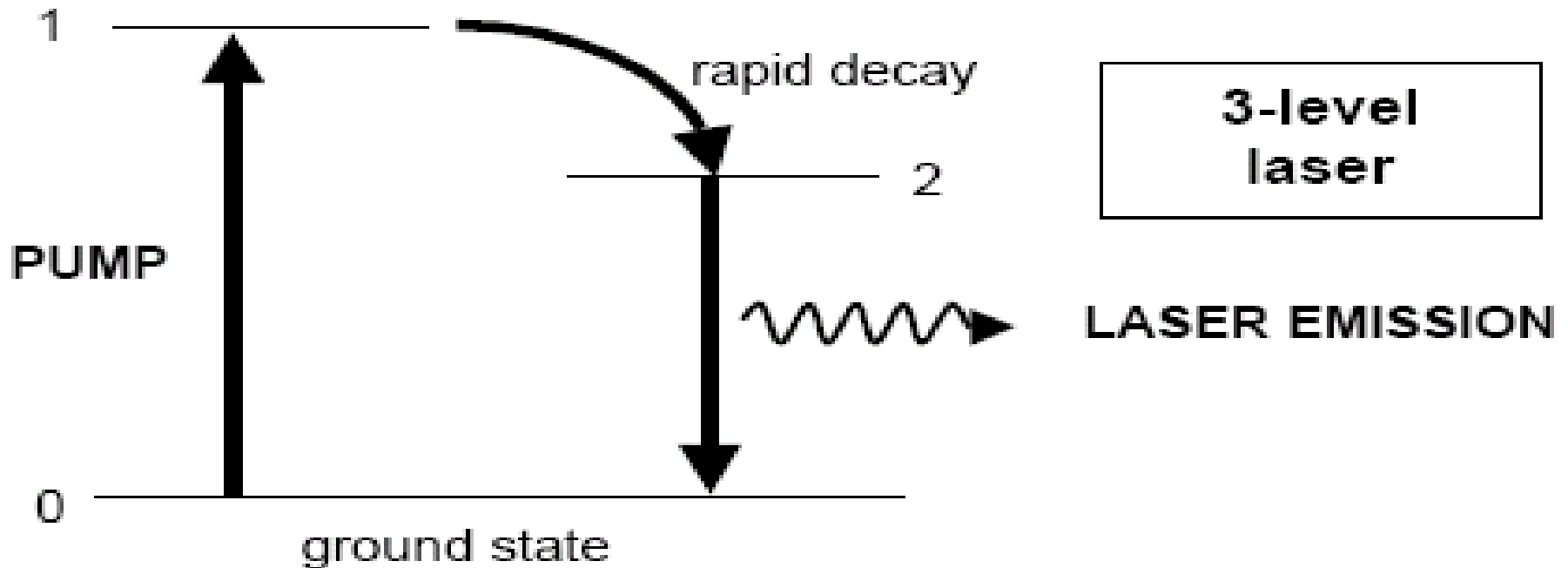
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Dr. Mohammed Abdullah Jassim

Laser level systems

Triple-Level System:

It consists of three energy levels: the ground level, which represents the **lower laser level (LLL)**, the excited level, which represents the **upper laser level (ULL)**, and the metastable or intermediate level (**ULL**).



The **ground level** is the same as the **lower laser level**. Half the number of atoms or molecules from the ground level must be pumped to the upper level to achieve the inverted distribution, so a very high pumping energy is required. The metastable level is not chosen for the inverted distribution process because it cannot store as many excited atoms or molecules as the upper laser level, which is very wide

The laser output power (PL) is calculated as follows:

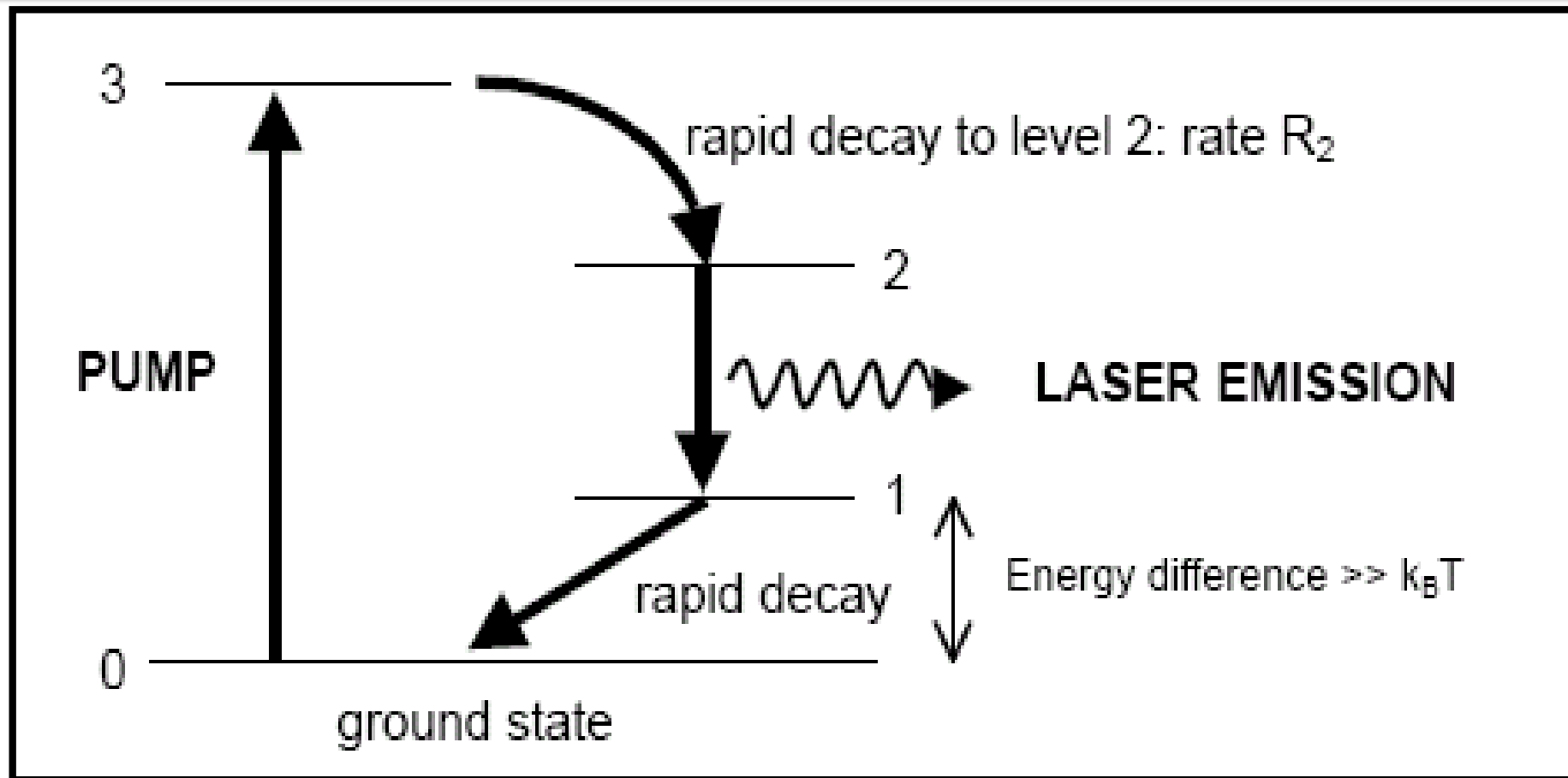
$$P_L = h\nu(w_p\beta N_1 - A_{21}N_2)$$

Where $h\nu$ is the energy of the emitted photon (laser), w_p is the pumping rate of atoms or molecules to the upper level, β is the **efficiency** of the N_2 level, and A_{21} is the spontaneous emission rate from the metastable level to the ground level.

Four-Level System:

It consists of four energy levels:

the ground level, the lower laser level (LLL), the excited level, and the upper laser level (ULL)



The ground **level** is not the same as the lower laser level , so we don't need a very powerful pumping source to achieve the inverted distribution. Most materials used for laser generation are four-level systems.

The laser output power (PL) is calculated as follows:

$$P_L = h \nu \Delta N_c w_L = h \nu \Delta N_c \left(\frac{P_2'}{\Delta N_c} - w_{21} \right)$$

Where $h\nu$ is the energy of the emitted photon (laser), ΔN_c is the value of the inverse distribution, w_L is the pumping or descent rate of atoms or molecules from the upper laser level to the lower laser level, and P_2 is the effective pumping power, calculated as follows:

$$P_2' = P_2 \left[1 - \left(\frac{w_2}{w_{10}} \right) \left(1 + \frac{P_1}{P_2} \right) \right]$$

If the number of atoms or molecules N_2 is greater than N_1 by one, this means that the inverse distribution between levels E_2 and E_1 has occurred.

Conclusion

Three-level system

- 1-It consists of three levels
- 2-It requires pumping half the number of atoms from the ground state to the excited state to obtain the inverted distribution
- 3-The ground plane is the same as the lower laser plane
- 4-Laser action occurs between E2 and E1
- 5-Requires a high pumping power source.
- 6-The **lifespan** of the E3 level is very small.
- 7- E2 level has a long **lifespan**
- 8-Rapid transition occurs between E3 and E2

9-Less efficient than the four-level system

10-The power of the laser depends on:

- 1-The **frequency** of the laser beam (ν)
2. The **probability** of stimulated transition (w_p)
3. The level **efficiency** (β) E_2
4. The probability of **spontaneous** emission A_{21}
5. The **number** of **atoms** in the N_1 and N_2 levels

11-Under **thermal equilibrium**, N_2 is **very small** and can be neglected

12-There is **no transition** between E_2 and the **ground** plane.

Conclusion

Four-level system

1-It consists of four levels.

2-It requires pumping a few atoms from the ground plane to the excited plane to obtain the inverted distribution

3-The ground plane is not the lower laser plane

4-Laser action occurs between E3 and E2.

5-No high pumping power source required

6-The lifespan of the E3 level is very small

7-The lifetime of the E2 level is relatively long.

8-Rapid transition occurs between E3 and E2

9-Higher efficiency than the three-level system

10-The laser power depends on:

- 1**-The frequency of the laser beam (ν)
- 2**. The probability of the stimulated transition (w_p)
- 3**. The effective pumping rate ($P'2$)
- 4**. The difference in the number of atoms in the **N1** and **N2** levels (ΔN_c)

11-Under thermal equilibrium, **N1** and **N2** are so small that they can be neglected

12-There is a transition between level **A1** and the ground level

Thanks for
listening.