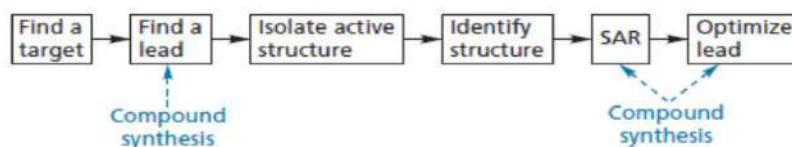


# Combinatorial and parallel synthesis

The full set of compounds produced by Combinatorial and parallel synthesis is called a **compound library**

*Lec. 7*

## Combinatorial and parallel synthesis in medicinal chemistry projects



- The procedures used in combinatorial synthesis are designed to produce **mixtures of different compounds** within each reaction vessel, whereas those used in parallel synthesis produce a **single product** in each vessel.

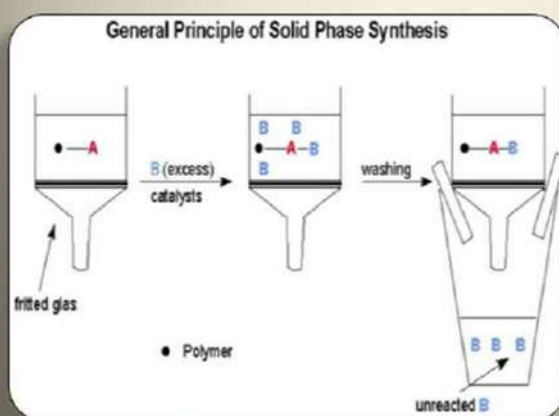
## Solid phase techniques

➤ Solid phase techniques can be used to carry out reactions where the starting material is linked to a solid support, such as a resin bead.

### Advantages:

- excess reagents or unbound by-products from each reaction can be easily removed by washing the resin.
- large excesses of reagents can be used to drive the reactions to completion (greater than 99%).
- intermediates in a reaction sequence are bound to the bead and do not need to be purified.

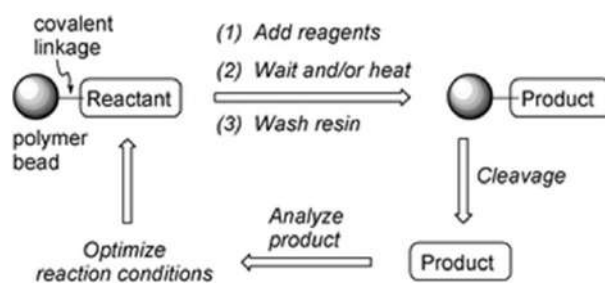
### SOLID PHASE TECHNIQUE :



❖ The solid support  
e.g. Cross-linked polystyrene Bead

❖ The anchor / linker  
e.g. Polystyrene resin ,  
Tentagel resin ,  
Polyacrylamide resin,  
Glass & ceramic beads .

## Solid phase techniques



## Solid phase techniques

### Advantages:

- the polymeric support can be regenerated and reused.
- automation is possible.
- if a combinatorial synthesis is being carried out, a range of different starting materials can be bound to separate beads. The individual beads can be separated at the end of the experiment to give individual products.

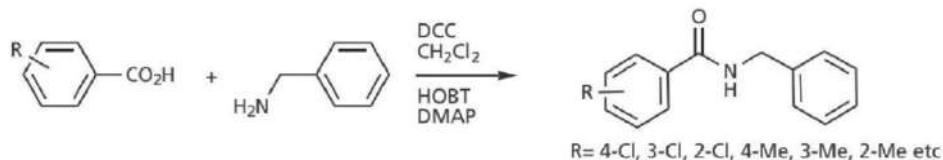
## The essential requirements for solid phase synthesis

- a cross-linked insoluble **polymeric support** which is **inert** to the synthetic conditions (e.g. a resin bead);
- an **anchor or linker** covalently linked to the resin—the anchor has a reactive functional group that can be used to attach a substrate.
- a **bond** linking the substrate to the linker, which will be stable to the reaction conditions used in the synthesis;
- a **means of cleaving** the product or the intermediates from the linker.
- **protecting groups** for functional groups not involved in the synthetic route

## Parallel synthesis

- a reaction is carried out in a series of wells such that each well contains a **single** product. This method is a ‘**quality** rather than **quantity**’
- often used for focused **lead optimization** studies.
- typical medicinal chemist may synthesize **one** or **two** new entities a week.
- With **parallel synthesis**, that **same researcher** can synthesize a **dozen** or more pure molecules.
- can be carried out on **solid phase** and also be carried out **in solution** “**solution phase organic synthesis (SPOS)**”
- many techniques to facilitate SPOS. Eg. Amide synthesis.

eg. Amide synthesis



Conventionally, a work-up procedure involves:

1. Washing the organic solution with aqueous **acid**.
2. Separate the two layers and organic layer is washed with an aqueous **base**.
3. Separate the two layers, and then the organic layer is treated with a **drying agent** such as magnesium sulphate.
4. The drying agent is **filtered** off and then the solvent is removed to afford the crude amide.
5. **Purification** then has to be carried out by **crystallization** or **chromatography**.

In **parallel synthesis** variety of useful **techniques** can also be used to minimize the work-up procedure:

1. **Equipment miniaturization**: This enable one to perform up to 24 reactions followed by 24 simultaneous evaporations on a normal heater stirrer unit.
2. Multiple parallel or sequential **automated chromatography units**
3. **Microwave** reactors can dramatically **speed up** reaction times.

## Combinatorial synthesis

- In combinatorial synthesis, **mixtures** of compounds are deliberately produced in each reaction vessel,
- The structures in each reaction vessel of a combinatorial synthesis are **not separated** and **purified**, but are tested for biological activity **as a whole**.
- Active mix .....**one or more cpd active** or **false positive**.
- Inactive mix .... then there is no need to continue studies on that mixture and it is stored.
- Overall, there is an **economy of effort**, as a negative result for a mixture of 100 compounds saves the effort of **synthesizing**, **purifying**, and **identifying** each component of that mixture.

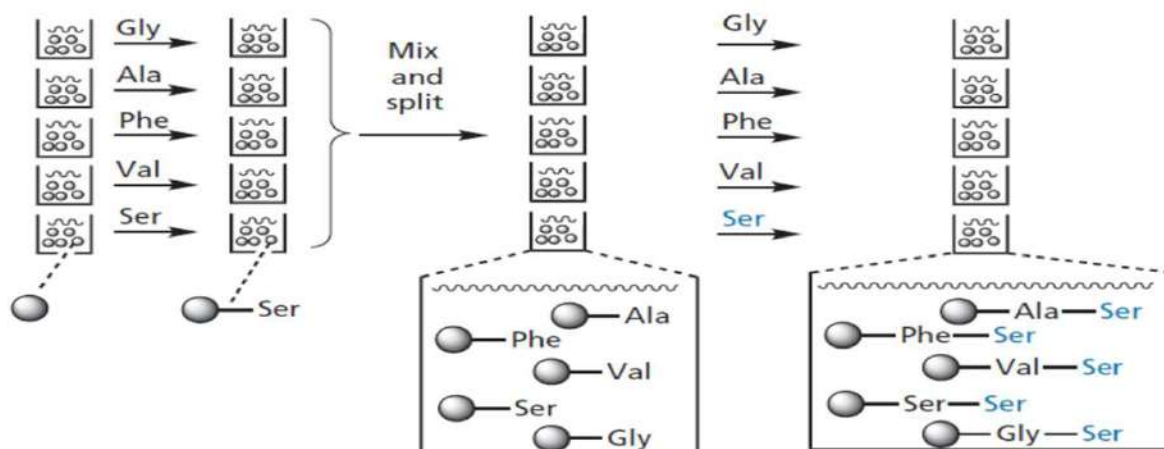
## The mix and split method in combinatorial synthesis

- Mix and split strategy used to **minimize the effort** involved and to **maximize the number** of different structures obtained.
- **NO. of possible dipeptides.**

Gly	25 separate procedures →	Gly-Gly	Ala-Gly	Phe-Gly	Val-Gly	Ser-Gly
Ala		Gly-Ala	Ala-Ala	Phe-Ala	Val-Ala	Ser-Ala
Phe		Gly-Phe	Ala-Phe	Phe-Phe	Val-Phe	Ser-Phe
Val		Gly-Val	Ala-Val	Phe-Val	Val-Val	Ser-Val
Ser		Gly-Ser	Ala-Ser	Phe-Ser	Val-Ser	Ser-Ser

## By using mix and split method:

Each individual bead may contain a **large number** of molecules, but all the molecules on that bead are **identical**

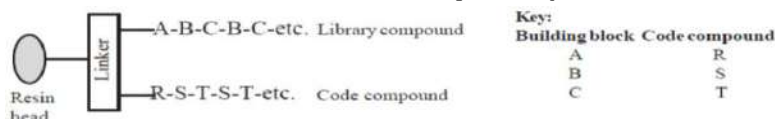


**FIGURE 16.29** Synthesis of five different dipeptides using the mix and split strategy.

## Structure determination of the active compound(s)

### Tagging:

two molecules are built up on the same bead. One of these is the intended structure, the other is a molecular tag (usually a peptide or oligonucleotide) which will act as a code for each step of the synthesis.



Compounds used for tagging must satisfy a number of **criteria**:

- (1) The **concentration** of the tag should be just sufficient for its analysis, that is, the majority of the linkers should be occupied by the combinatorial synthesis.
- (2) The **tagging reaction** must take place under **conditions** that are compatible with those used for the synthesis of the library compound.
- (3) It must be possible to **separate** the tag from the library compound.
- (4) **Analysis** of the tag should be **rapid** and **accurate** using methods that could be automated.

**Table 5.2** The use of oligonucleotides to encode amino acids in peptide synthesis

Amino acid	Structure	Oligonucleotide code
Glycine (Gly)	$\text{NH}_2$ $\text{CH}_2\text{COOH}$	CACATG
Methionine (Met)	$\text{NH}_2$ $\text{CH}_3\text{SCH}_2\text{CH}_2\text{CHCOOH}$	ACGGTA

