

# Quantum Mechanics in Medicine

## Lecture 6

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For Third-year Students

# Outline

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# Time-Independent Schrödinger Equation

In many physically interesting cases, the potential energy  $V$  is time-independent:  $V = V(\mathbf{r})$ .

In such cases, we may apply the **method of separation of variables**.

Assume that the total wave function can be written as a product:

$$\Psi(\mathbf{r}, t) = \psi(\mathbf{r})f(t).$$

Substituting into the Schrödinger equation yields separate equations for space and time, each equal to a constant.

# Derivation and Energy Constant

Starting from the time-dependent Schrödinger equation:

$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \nabla^2 \Psi + V(\mathbf{r}) \Psi$$

and using the separable form  $\Psi(\mathbf{r}, t) = \psi(\mathbf{r})f(t)$ , we obtain:

$$i\hbar \psi(\mathbf{r}) \frac{df(t)}{dt} = f(t) \left[ -\frac{\hbar^2}{2m} \nabla^2 + V(\mathbf{r}) \right] \psi(\mathbf{r})$$

which separates the time and spatial parts. Dividing by  $\psi(\mathbf{r})f(t)$  gives

$$\frac{1}{f} \frac{df}{dt} = -\frac{iE}{\hbar}, \quad \left[ -\frac{\hbar^2}{2m} \nabla^2 + V(\mathbf{r}) \right] \psi = E\psi.$$

Integrating yields  $f(t) = e^{-iEt/\hbar}$ ; the second equation is the **time-independent Schrödinger equation (TISE)**.

# Eigenvalue Equation Form

$$\hat{H}\psi(\mathbf{r}) = E\psi(\mathbf{r}), \quad \hat{H} = -\frac{\hbar^2}{2m}\nabla^2 + V(\mathbf{r}).$$

$\hat{H}$  acting on  $\psi$  multiplies it by  $E$ ; such an equation is an **eigenvalue equation**.  
 $\psi$  is an **eigenfunction**,  $E$  an **eigenvalue**.

The set of all  $E$  is the **eigenvalue spectrum**.

The problem of solving the Schrödinger equation reduces to finding eigenvalues and eigenfunctions of  $\hat{H}$ .

## Degeneracy

Sometimes more than one linearly independent eigenfunction corresponds to the same eigenvalue.

Such an eigenvalue is **degenerate**.

If there are  $k$  independent eigenfunctions for the same  $E$ , the eigenvalue is  **$k$ -fold degenerate**.

Any linear combination of degenerate eigenfunctions is also an eigenfunction:

$$\Psi = c_1 \Psi_1 + c_2 \Psi_2 + \cdots + c_k \Psi_k.$$

# Reality of Eigenvalues

Let  $E$  be the eigenvalue corresponding to eigenfunction  $\Psi$ :

$$\hat{H}\Psi = E\Psi.$$

The Hamiltonian  $\hat{H}$  is a **Hermitian operator**.

Therefore, all energy eigenvalues are **real**.

# Stationary States

For a separable wavefunction, the probability density is

$$|\Psi(\mathbf{r}, t)|^2 = |\psi(\mathbf{r})|^2 |f(t)|^2 = |\psi(\mathbf{r})|^2,$$

which is independent of time.

These are called **stationary states**.

The expectation value of total energy equals the eigenvalue  $E$  for all time if the wavefunction is normalized.

# Orthogonality and Orthonormality

Eigenfunctions corresponding to distinct eigenvalues are **orthogonal**.

If normalized, we combine orthogonality with normalization:

$$\int \psi_k^* \psi_n \, d\tau = \begin{cases} 1, & k = n, \\ 0, & k \neq n. \end{cases}$$

This is the **orthonormality condition**.

## Parity in One Dimension

Suppose the potential is symmetric:  $V(-x) = V(x)$ .

Under the reflection  $x \rightarrow -x$  (parity operation), Schrödinger's equation is invariant.

If  $\psi(x)$  is an eigenfunction,  $\psi(-x)$  is another eigenfunction with the same eigenvalue  $E$ .

## Even and Odd Wavefunctions

If the eigenvalue is nondegenerate,  $\psi(-x)$  and  $\psi(x)$  differ only by a constant. Hence, eigenfunctions can be classified as:

$$\psi(-x) = +\psi(x) \quad (\text{even}), \quad \psi(-x) = -\psi(x) \quad (\text{odd}).$$

**Q1.** For a time-independent potential  $V(\mathbf{r})$ , the wavefunction can be written as:

- A.  $\Psi(\mathbf{r}, t) = \psi(\mathbf{r}) + f(t)$
- B.  $\Psi(\mathbf{r}, t) = \psi(\mathbf{r})f(t)$
- C.  $\Psi(\mathbf{r}, t) = \psi(\mathbf{r})t$
- D.  $\Psi(\mathbf{r}, t) = \psi(\mathbf{r}) - f(t)$

**Q2.** The separation constant appearing after separation of variables equals:

- A. Momentum
- B. Normalization constant
- C. Energy  $E$
- D. Parity

**Q3.** The time part of an energy eigenfunction is proportional to:

- A.  $e^{-iEt/\hbar}$
- B.  $e^{+iEt/\hbar}$
- C.  $e^{-Et/\hbar}$
- D.  $e^{+Et/\hbar}$

**Q4.** The time-independent Schrödinger equation is expressed as:

- A.  $\hat{H}\psi = E\psi$
- B.  $\hat{p}\psi = i\hbar\partial_x\psi$
- C.  $\nabla^2\psi = 0$
- D.  $\hat{H}\psi = iE$

**Q5.** The complete set of all eigenvalues of  $\hat{H}$  is the:

- A. State space
- B. Eigenvalue spectrum
- C. Configuration space
- D. Momentum space

**Q6.** More than one independent eigenfunction with the same  $E$  means the level is:

- A. Normalized
- B. Stationary
- C. Degenerate
- D. Orthogonal

**Q7.** A linear combination of degenerate eigenfunctions is:

- A. Not an eigenfunction
- B. An eigenfunction with the same eigenvalue
- C. An eigenfunction with a different eigenvalue
- D. Only approximately an eigenfunction

**Q8.** Hermiticity of  $\hat{H}$  ensures that:

- A. Energy eigenvalues are always complex
- B. Energy eigenvalues are always real
- C. Energy eigenvalues are imaginary
- D. Energy eigenvalues are arbitrary

**Q9.** For  $\Psi = \psi e^{-iEt/\hbar}$ , the probability density is:

- A. Time-dependent
- B. Constant in time
- C. Exponentially decaying
- D. Oscillatory

**Q10.** The expectation value of  $\hat{H}$  for a normalized eigenstate equals:

- A.  $\hbar$
- B. 0
- C. The eigenvalue  $E$
- D. Only kinetic energy

**Q11.** Eigenfunctions of distinct eigenvalues are:

- A. Orthogonal
- B. Parallel
- C. Equal
- D. Arbitrary

**Q12.** The orthonormality condition is:

- A.  $\int \psi_k^* \psi_n d\tau = \delta_{kn}$
- B.  $\int \psi_k \psi_n d\tau = 1$
- C.  $\int |\psi_n|^2 d\tau = 0$
- D.  $\psi_k = \psi_n$

**Q13.** For  $V(-x) = V(x)$ , the system is symmetric under:

- A. Time reversal
- B. Parity operation
- C. Translation
- D. Scaling

**Q14.** Nondegenerate energy levels in even potentials correspond to eigenfunctions that are:

- A. Even or odd
- B. Only even
- C. Only odd
- D. Neither even nor odd

## MCQs (15–16)

**Q15.** The parity relation for even and odd functions is:

- A.  $\psi(-x) = +\psi(x)$  or  $\psi(-x) = -\psi(x)$
- B.  $\psi(-x) = i\psi(x)$
- C.  $\psi(-x) = \psi^*(x)$
- D.  $\psi(-x) = 0$

**Q16.** The one-dimensional Schrödinger equation is:

- A.  $-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + V(x)\psi = E\psi$
- B.  $i\hbar \frac{\partial\psi}{\partial t} = \hat{H}\psi$
- C.  $\hat{p}\psi = -i\hbar \frac{d\psi}{dx}$
- D.  $\nabla^2\psi = 0$

## MCQs (17–18)

**Q17.** If  $\psi_1, \psi_2$  are orthogonal eigenfunctions with the same  $E$ ,  $c_1\psi_1 + c_2\psi_2$  is:

- A. Not normalizable
- B. An eigenfunction with eigenvalue  $E$
- C. An eigenfunction with eigenvalue  $E_1 + E_2$
- D. Purely imaginary

**Q18.** The 3D Hamiltonian operator is:

- A.  $\hat{H} = -\frac{\hbar^2}{2m}\nabla^2 + V(\mathbf{r})$
- B.  $\hat{H} = i\hbar\frac{\partial}{\partial t}$
- C.  $\hat{H} = \nabla + V$
- D.  $\hat{H} = \mathbf{r}$

**Q19.** Hermiticity of  $\hat{H}$  implies:

- A. Real measurable energies
- B. Imaginary eigenvalues
- C. Infinite degeneracy
- D. Random phases

**Q20.** Stationary states have:

- A. Constant probability density
- B. Decaying probability density
- C. Increasing momentum
- D. Nonconserved energy