



**Al-Mustaqbal University**  
**College of Science**



# **Scientific Thinking and Research Skills**

**Third Year Students / 1<sup>st</sup> Semester**

**INTRODUCTION TO RESEARCH METHODOLOGY, *continued***

**2025-2026**

***Biochemistry Department***

***By***

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# What is Methodology?

A set of procedures for the purpose of answering a research question(s) that describes:

- ☐ **How** to study (object /participants ) will be selected?
- ☐ **How** you will analyze the data?
- ☐ **How** and **when** you will collect data from your work or participants?



# Defining Research

## Methodology



**Methodology means the systematic study of the  
methods, principles, procedures, and rules  
applied within a discipline**



# Research Methodology

1. Similar to an instruction booklet, whereby the instruments and procedures used in the research are described.
2. It should be clearly written so that other researchers who are interested in conducting research in the same area can exactly follow the procedures you took in conducting the research. .

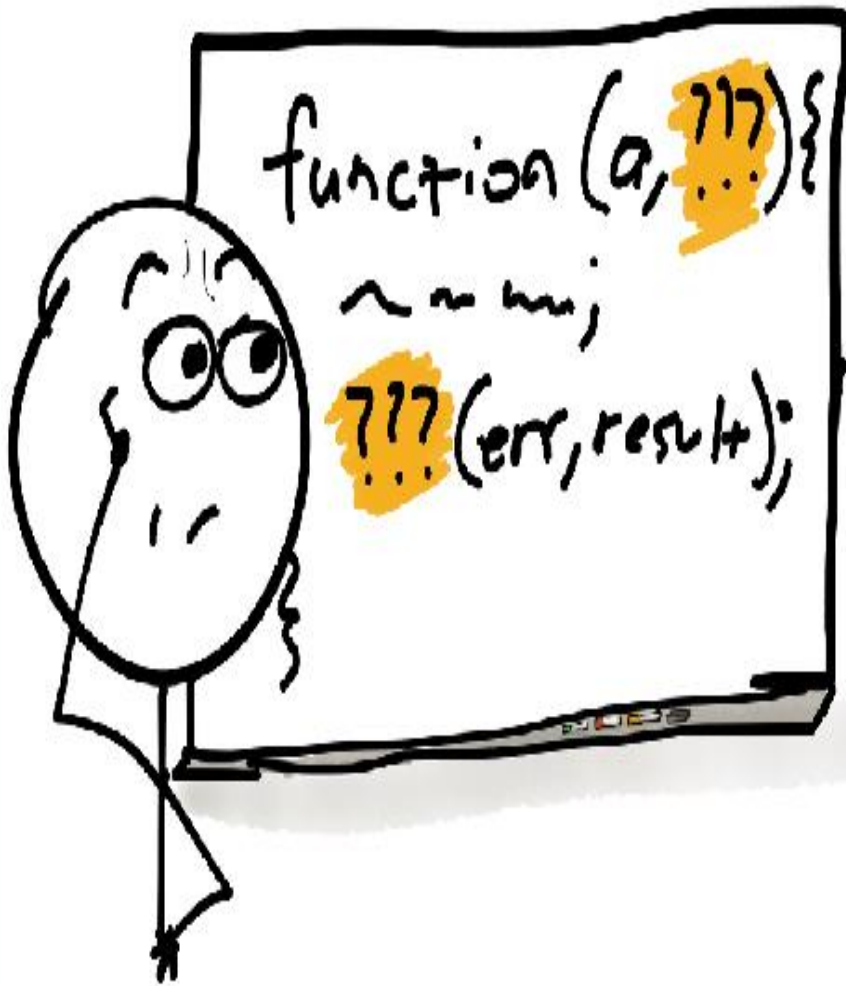


# Main Elements of Research Methodology

- ❖ *Purpose or objectives of the research.*
- ❖ *Short description of the issue or problem to be investigated.*
- ❖ *Data collection methods.*
- ❖ *Data analysis methods.*



Ask your self, What the scientific meaning for these pictures?



OBSERVATION



QUESTION



RESEARCH AND HYPOTHESIS



EXPERIMENT



ANALYSIS AND CONCLUSION



## Steps of the Scientific Method



Observation

Which type of fertilizer works the best?



Question



Hypothesis



Results



Conclusion



# Research Methodology Sections

1. Introduction.
2. Description of Samples/Subjects.
3. Research Procedure.
4. Data Analysis and Conclusion.
5. Description of Instruments Used.





# Students Exercise

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Full Length Article

## Morphological driven photocatalytic activity of ZnO nanostructures



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### ABSTRACT

Using a simple combination of pulse laser ablation in liquid and hydrothermal (PLAL-H) approaches, we control the morphology of ZnO nanostructures (ZNSs) to determine the feasibility of their photocatalytic efficacy. These ZNSs are deposited on Si (100) substrates and two different morphologies are achieved. In this synergistic approach, PLAL synthesized NSs are used as a nutrient solution with different pH for further hydrothermal treatment at 110 °C under varying growth time (5, 30 and 60 min). Surface morphology, structure, composition, and optical characteristics of the prepared ZNSs are determined using FESEM, XRD, FTIR and Photoluminescence (PL) and UV–vis absorption measurements. The morphology revealed remarkable transformation from nanorods (NRs)/nanoflowers (NFs) (at pH 7.6) to nanoparticles (NPs)-like (at pH 10.5) structure. XRD patterns showed better polycrystallinity for NPs with enlarged band gap than NR/NF-like structures. Both PL and UV–vis spectral analysis of ZNPs exhibited higher surface area and deep level defects density dependent morphology, where the nutrient pH and growth time variation are found to play a significant role towards structural evolution. Furthermore, the photocatalytic activities of such ZNSs are evaluated via sunlight driven photo-degradation of methylene blue (MB) dye



## 1. Introduction

## 2. Description of Samples/Subjects

## 3. Research Procedure

## 4. Data Analysis

## 5. Conclusion

## 6. Description of Instruments Used

### 2. Experimental

#### 2.1. Samples preparation

Different types of ZNSSs-based films are synthesized under moderate growth conditions by combining PLAL method with hydrothermal method which is indigenously developed [40]. The growth process is comprised of two steps, where the colloidal ZnO nanomaterial is first prepared via PLAL method which is then used as nutrients for further hydrothermal treatment [40,41]. An analytical grade metallic Zn plate of 1 mm thick (purity 99.9% from Sigma Aldrich) is used as target material, which is cleaned in an ultrasonic ethanol bath for 30 min followed by thorough rinsing with deionized water to eliminate the pollutants. Subsequently, this Zn target is fixed at the bottom of a glass vessel filled with 40 mL of distilled water before being ablated using a pulsed Nd:YAG laser source (ablation wavelength of 532 nm, ablation time of 6 min, laser fluence of  $12 \text{ mJ/cm}^2$ , pulse duration of 8 ns and repetition rate of 10 Hz). The solution is continuously stirred during the ablation process to reduce the aggregation of ZNPs, which acts as an obstacle to the pulse irradiation path. At the end of laser ablation, a colloidal solution-based ZnO nanomaterial is obtained. Next, such PLAL generated nanomaterial is subjected to hydrothermal treatment for achieving ZNSSs with diverse morphologies. The hydrothermal treatment is performed at  $110^\circ\text{C}$  with varying nutrient pH in the range of 7.6–10.5 and growth time of 5, 30 and 60 min. Two types of ZNSSs-based films including NRs/NFs and NPs-like morphologies are successfully grown on Si (100) substrates. The pH value of the nutrient solution (colloidal solution-based ZnO nanomaterial) is adjusted by adding the aqueous  $\text{NH}_3$ . The hydrothermal treatment is carried out to enhance the coverage and the growth rate of ZNSSs-based films deposited on the Si substrates. Eventually, the prepared samples are cooled down to room temperature, followed by rinsing with deionized water and drying at  $80^\circ\text{C}$  for 60 min.

#### 2.2. Characterization

The surface morphology and the structural properties of as-



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## 2.2. Characterization

The surface morphology and the structural properties of as-grown ZNSs are determined using FESEM (Hitachi 4100 – SU8020) and XRD (Rigaku, 2100) measurements. The FTIR spectra in the wavenumber range of  $400\text{--}4000\text{ cm}^{-1}$  are recorded using a PerkinElmer spectrum GX spectrophotometer. PL spectrum is captured using Perkin-Elmer, Luminescence LS55 spectrophotometer equipped with a Xenon lamp as excitation source (wavelength of 320 nm). The UV-vis absorption spectra in the wavelength range of 200–800 nm are measured using a PerkinElmer Lambda-2 spectrometer. All the characterizations are performed at room temperature (RT).

## 2.3. Photocatalytic activity evaluation

The photocatalytic activities of synthesized ZNSs (NPs, and NRs/NFs) under solar irradiation are evaluated via the photo-degradation of MB dye in an aqueous solution possess pH value (7.4). An irradiated area of  $4.5\text{ cm}^2$  is selected for both types of as



Thanks

Questions ?





*How To Do a Research*

*How to write your thesis*



# ***CHAPTER 1***

1. What is the Introduction?
2. Problem Statement?
3. Purpose of the Study?
4. Objectives of the Study?
5. Research Questions?
6. Significance of the Study?
7. Scope of the Study ?